









Synthesis and Application of Castor Oil based MacroPhotoinitiators and their Application for Biobased Polymer Coatings and Gold Nanocomposites

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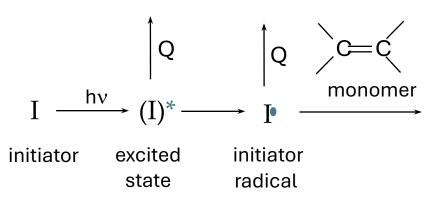








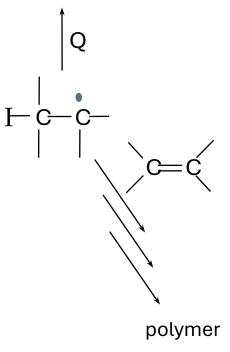






• efficient generation of the triplet state

• fast radical generation



high addition rate constant







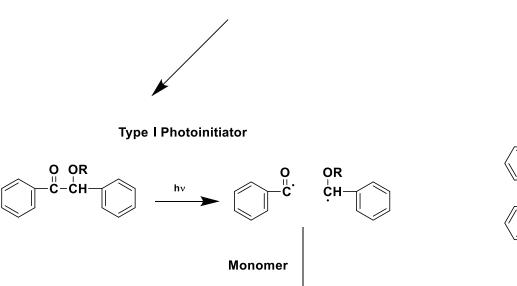




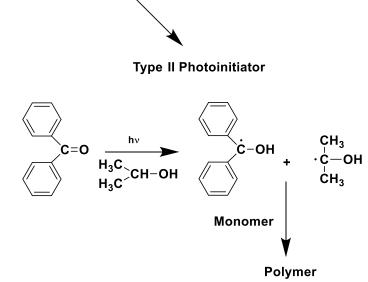




### **Photoinitiator**



Polymer







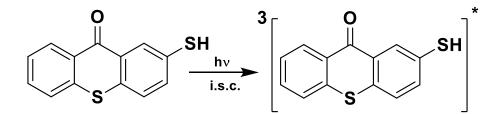


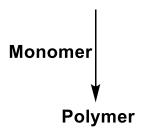


























#### Oils from Renewable Sources

Soybean Oil Canola Oil







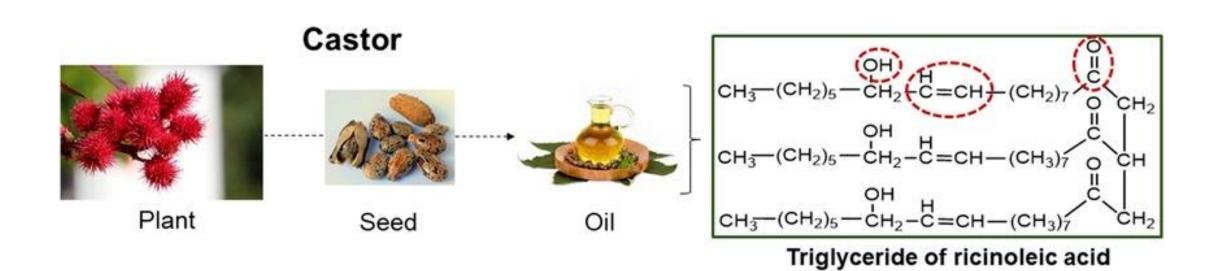








#### Castor Oil





Materials







Polyricinoleic acid

# **Results and Discussion**

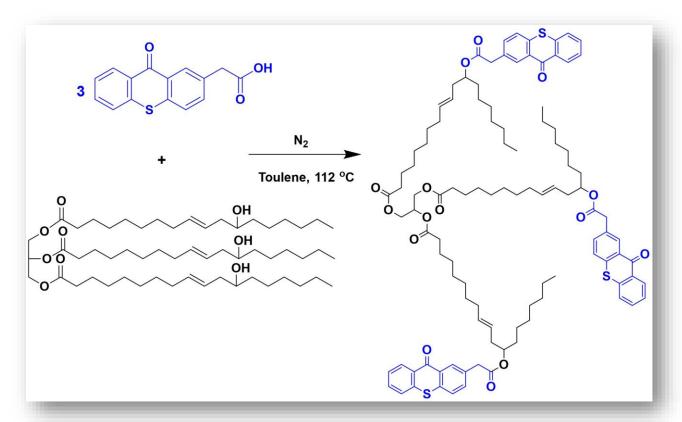








# Esterification reaction of Castor Oil with TXCH<sub>2</sub>COOH (PICO I)



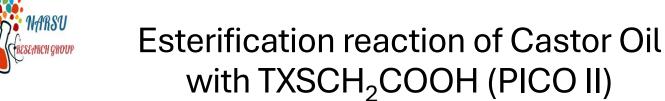






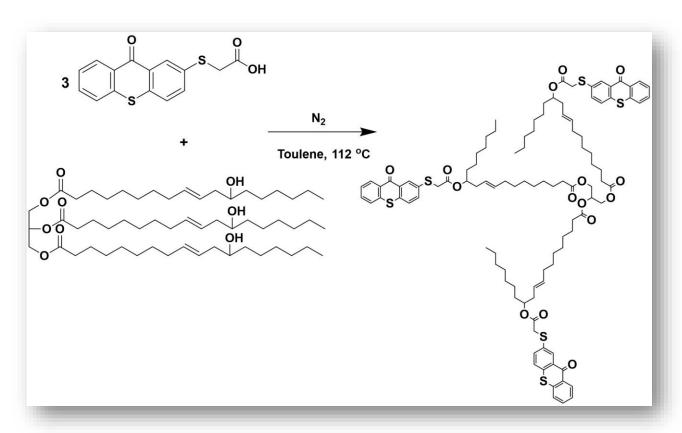


















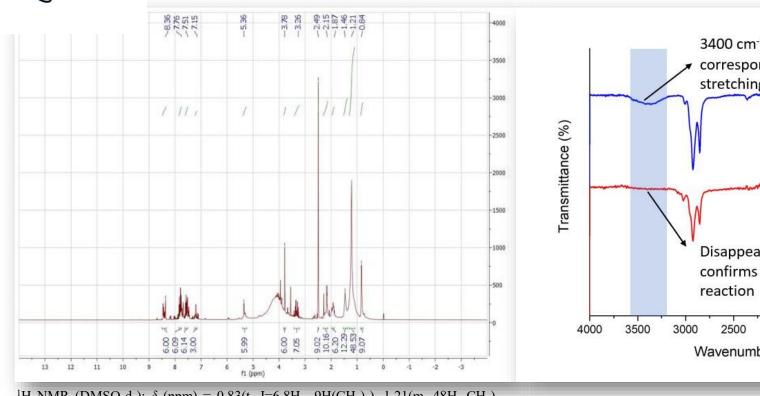


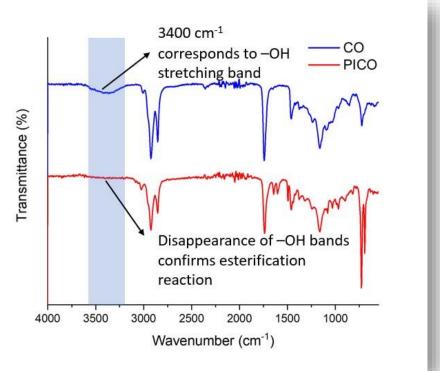


## Characterization by NMR and FTIR









<sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  (ppm) = 0.83(t, J=6.8H<sub>2</sub>, 9H(CH<sub>3</sub>)<sub>3</sub>), 1.21(m, 48H, CH<sub>2</sub>), 1.46(br s, 12H, CH<sub>2</sub>), 1.87-1.99 (m, 6H, CH<sub>2</sub>), 2.15-2.20 (m, 7H, CH and CH<sub>2</sub>), 2.49 (dt, J=1.8; 3.6H<sub>2</sub>, 9H, CH and CH<sub>2</sub>), 3.26-3.41 (m, 7H, CH and CH<sub>2</sub>), 3.78 (s, 6H, (CH<sub>2</sub>)<sub>3</sub>), 5.36 (s, 6H, (CH<sub>2</sub>)<sub>3</sub>), 7.15-7.24 (m, 3H, aryl-H), 7.51-7.62 (m, 6H, aryl-H), 7.76-7.84 (m, 6H, aryl-H), 8.36-8.47 (m, 6H, aryl-H)









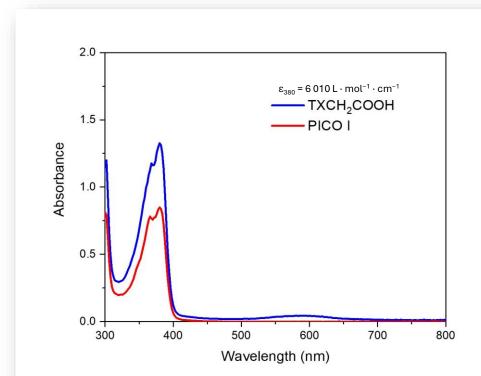




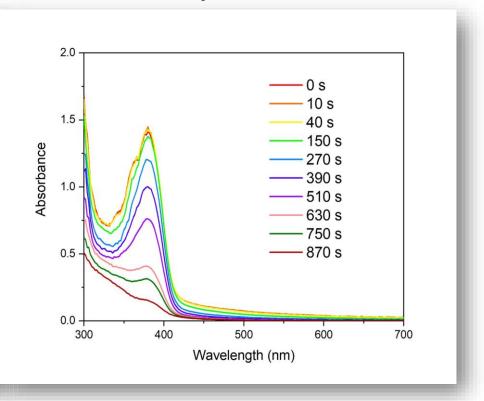


#### **UV-Vis Studies**

#### UV-Vis studies for PICO I



#### Photolysis in EtOH











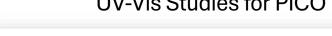


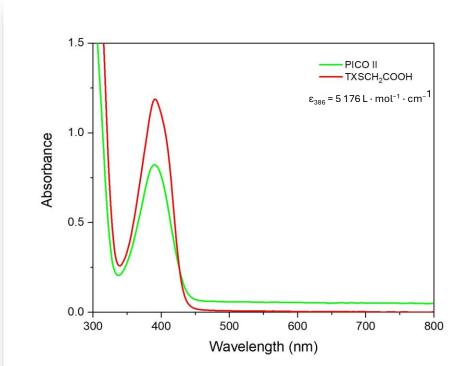
#### **UV-Vis Studies**



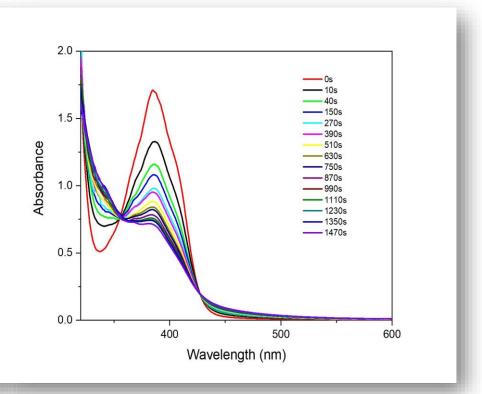


#### UV-Vis Studies for PICO II





#### Photolysis in DCM















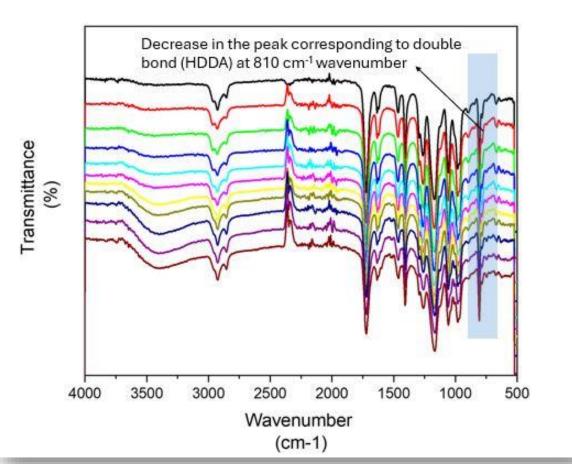


# Real-time FTIR studies(RT-FTIR)

HDDA 70% (w/w) + PICO2 25% (w/w) + NMDEA 5% (w/w) Conversion degree formula is reported below:

$$Conversion\% = \frac{A_0 - A_t}{A_0} \times 100$$









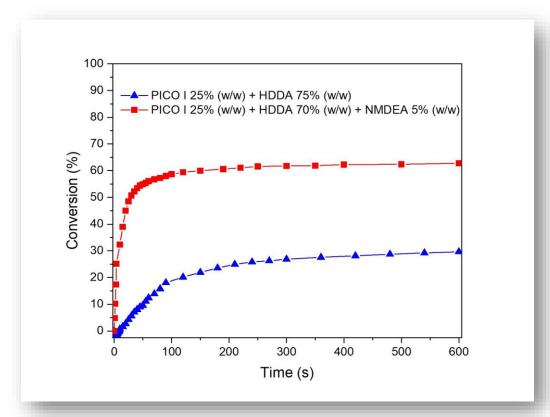


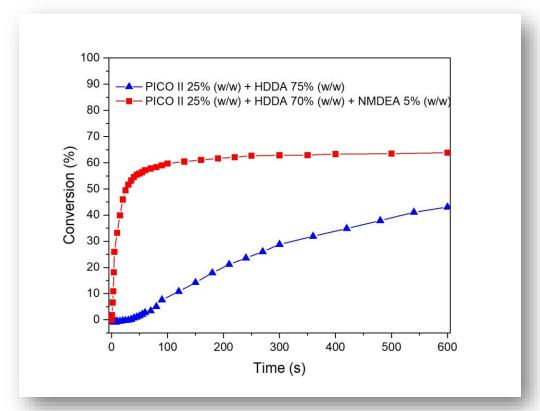






### Real-time FTIR studies (RT-FTIR)















## Photoinitiated polymerization of methyl methacrylate (MMA) by PICO I in DCM





Run	CO	MMA	[PICO I] 10mg/10ml	тхсн₂соон	NMDEA	DCM	AIR	$N_2$	Conversion %
1	-	+	+	-	-	-	+	5 <del>-</del> 5	2.12
2	2	+	+	2	2:	12	~~	+	4.44
3	-	+	+	-	+	+	+	0.00	3.11
4	2	+	+	2	+	+	2	+	4.54
5	:=	+	. <del></del>	-	-	+	-	6 <del>.0</del> 8	<del>-</del> 8
6	+	+	2	+	<u> </u>	<del>+</del>	+	100	2.99
7	+	+	<del></del>	+	-	+	·	+	2.31
8	+	+	2	+	+	2世3	+	12.5	1.54
9	+	+	S <del>4</del>	+	+	(1 <del>11</del> ))	·	+	1.50
10	+	+	2	20	<u> </u>	+	<u>125</u>	100	928

 $[MMA] = 4.68 \text{ mol} \cdot L^{-1}$ 

 $[CO] = 1x10^{-3} M$ 

[TXCH<sub>2</sub>COOH] = 1x10<sup>-3</sup> M

 $[NMDEA] = 1x10^{-2}M$ 

Irradiation time = 60 minutes











## Photoinitiated polymerization of methyl methacrylate (MMA) by PICO II in DCM





Run	CO	MMA	[PICO II] 10mg/10ml	TXSCH <sub>2</sub> COOH	NMDEA	DCM	AIR	$N_2$	Conversion %
1	_	+	+	949	- 2	(4)	+	46	2.16
2	-	+	+	-	. =	0.70	= 1	+	2.80
3	_	+	+	323	+	+	+	Δ)	3.18
4	-	+	+	-	+	+	-	+	2.00
5	144	+	2	((2)	2	+	929	<u> 28</u>	(2)
6	+	+	æ	+	5	+	+	=	3.15
7	+	+	<u>12</u>	+	2	+	343	+	2.24
8	+	+	æ	+	+	878	+	58	3.20
9	+	+	<u>12</u>	+	+	3343	323	+	3.22
10	+	+	æ	10 <del>7</del> 4	5.	+	8 <del>3</del> 8	75	5 <del>-</del> 7

 $[MMA] = 4.68 \text{ mol} \cdot L^{-1}$ 

 $[CO] = 1x10^{-3} M$ 

[TXSCH<sub>2</sub>COOH] = 1x10<sup>-3</sup> M

 $[NMDEA] = 1x10^{-2} M$ 

Irradiation time = 60 minutes







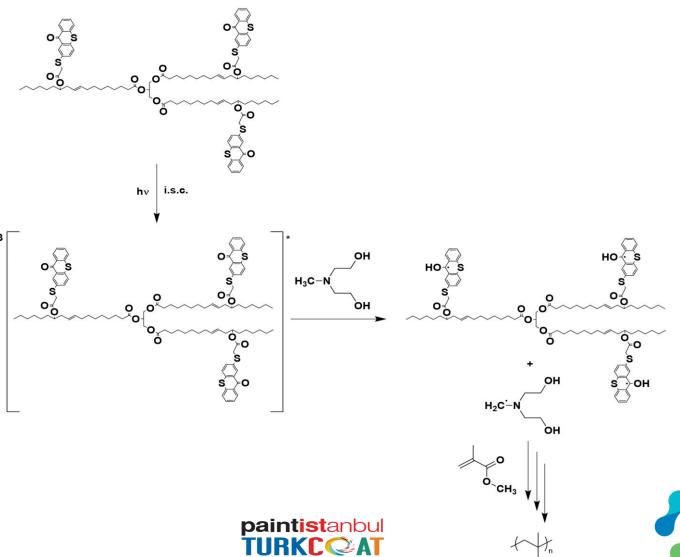




### Possible Initiation Mechanism of Thioxanthonated Castor Oil













PMMA polymer





### **UV-Curing Applications**





PICO I (25% w/w) + HDDA (75% w/w) formulation in absence (above) and in presence of NMDEA 5% w/w (below)















# **UV-Curing Applications**



PICO II (25% w/w) + HDDA (75% w/w) formulation in absence (above) and in presence of NMDEA 5% w/w (below)





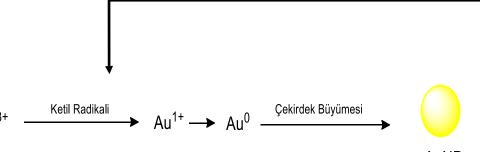












Ketyl radicals, which can be easily produced photochemically, are generally used to reduce metal salts.

Although ketyl radical is obtained effectively by Pinocol cleavage, they are not preferred because they are in competition for absorption with metal-containing substances (Low  $\lambda$ ).



A: Photoreduction

B: Norrish Type I cleavage

C: Pinacol Photocleavage



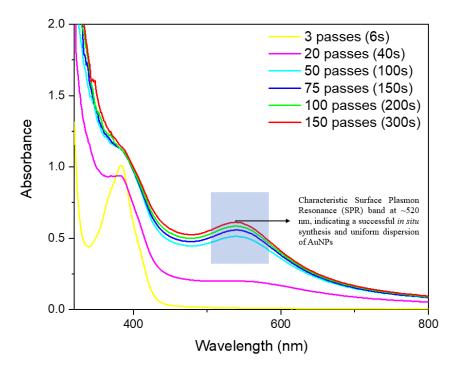






## In-situ Photochemical Synthesis of Nanocomposites Containing AuNPs

PICO II 50% (w/w) + NMDEA 5% (w/w) + HDDA 41% (w/w) + HAuCl<sub>4</sub> 4% (w/w)













# A biobased gold nanocomposite coating







PICO II (25% w/w) + HDDA (70% w/w) + NMDEA (5% w/w) (left) PICO II 50% (w/w) + NMDEA 5% (w/w) + HDDA 41% (w/w) + HAuCl<sub>4</sub> 4% (w/w) (right)







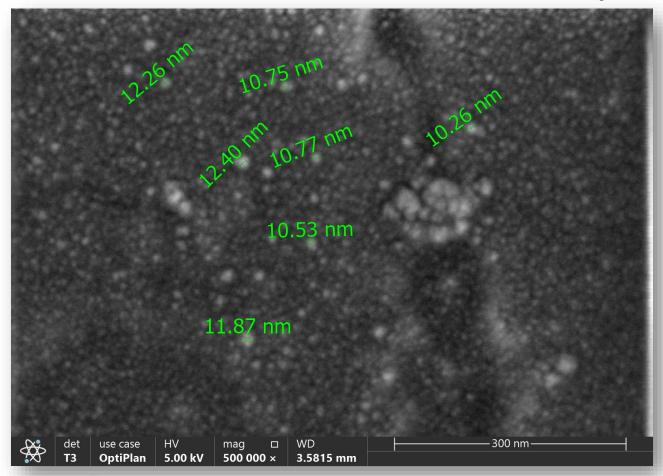


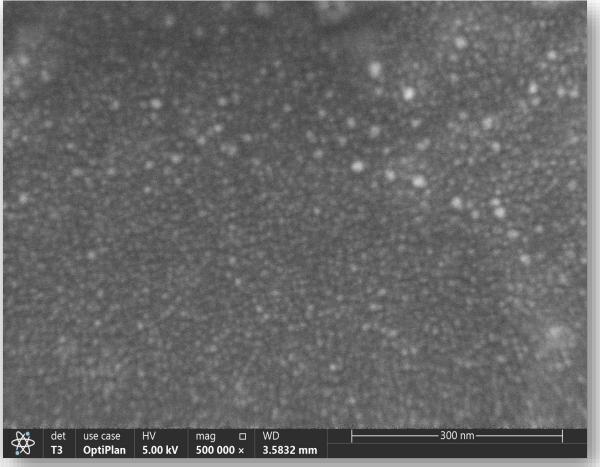


# SEM images of biobased nanocomposites





















#### Conclusion

- ✓ In this work, novel macrophotoinitiators were successfully synthesized by attaching thioxanthone (TX) moieties to a castor oil-derived backbone
- ✓ Combination of the macrophotoinitiator, HDDA, and NMDEA (N-methyldiethanolamine) led to the formation of highly transparent polymer films on glass slides, highlighting the effectiveness of the system in producing uniform and defect-free coatings.
- ✓ Incorporation of gold salt (HAuCl₄) into the formulation enabled the *in-situ* generation of gold nanoparticles during the photopolymerization process, resulting in gold nanocomposite materials











# Acknowledgement





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